Characterization of Plant Carotenoid Cyclases as Members of the Flavoprotein Family Functioning with No Net Redox Change1[W][OA]

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The later steps of carotenoid biosynthesis involve the formation of cyclic carotenoids. The reaction is catalyzed by lycopene β-cyclase (LCY-B), which converts lycopene into β-carotene, and by capsanthin-capsorubin synthase (CCS), which is mainly dedicated to the synthesis of κ-cyclic carotenoids (capsanthin and capsorubin) but also has LCY-B activity. Although the peptide sequences of plant LCY-Bs and CCS contain a putative dinucleotide-binding motif, it is believed that these two carotenoid cyclases proceed via protic activation and stabilization of resulting carbocation intermediates. Using pepper (Capsicum annuum) CCS as a prototypic carotenoid cyclase, we show that the monomeric protein contains one noncovalently bound flavin adenine dinucleotide (FAD) that is essential for enzyme activity only in the presence of NADPH, which functions as the FAD reductant. The reaction proceeds without transfer of hydrogen from the dinucleotide cofactors to β-carotene or capsanthin. Using site-directed mutagenesis, amino acids potentially involved in the protic activation were identified. Substitutions of alanine, lysine, and arginine for glutamate-295 in the conserved 293-FLEET-297 motif of pepper CCS or LCY-B abolish the formation of κ-carotene and κ-cyclic carotenoids. We also found that mutations of the equivalent glutamate-196 located in the 194-LIEDT-198 domain of structurally divergent bacterial LCY-B abolish the formation of β-carotene. The data herein reveal plant carotenoid cyclases to be novel enzymes that combine characteristics of non-metal-assisted terpene cyclases with those attributes typically found in flavoenzymes that catalyze reactions, with no net redox, such as type 2 isopentenyl diphosphate isomerase. Thus, FAD in its reduced form could be implicated in the stabilization of the carbocation intermediate.

Later steps of carotenoid biosynthesis involve the formation of diverse cyclic carotenoids. For example, β-carotene, the vitamin A precursor, is synthesized de novo by photosynthetic organisms, limited nonphototrophic bacteria and fungi, and also by aphids (Moran and Jarvik, 2010) according to a multistep pathway that ends with the cyclization of lycopene by lycopene β-cyclase (LCY-B). Similarly, in pepper (Capsicum annuum) chromoplasts, antheraxanthin and violaxanthin are converted into the κ-cyclic carotenoids capsanthin and capsorubin, respectively, by capsanthin-capsorubin synthase (CCS). In both cases, the proposed mechanism involves a concerted protic attack and stabilization of a transient carbocation without any net redox change (Camara, 1980; Bouvier et al., 1994; Britton, 1998). Several cDNAs for LCY-B have been cloned from bacteria (Misawa et al., 1990; Cunningham et al., 1994; Armstrong, 1997; Cunningham and Gantt, 2001), fungi (Verdoes et al., 1999; Velayos et al., 2000; Arrach et al., 2001), and plants (Hugueney et al., 1995; Ronen et al., 2000) using functional complementation. Information available from primary structures suggests that the cyclization of lycopene is catalyzed by homologous proteins in photosynthetic organisms (Cunningham et al., 1994; Maresca et al., 2007), by holomeric (Misawa et al., 1990) or heteromeric (Krubasik and Sandmann, 2000; Viveiros et al., 2000) proteins in nonphotosynthetic bacteria, and by holomeric, bifunctional proteins in fungi that combine the activities of phytoene synthase and lycopene cyclase (Verdoes et al., 1999; Velayos et al., 2000; Arrach et al., 2001). This structural

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diversity of LCY-Bs coupled to a lack of significant amino acid sequence identity between the lycopene cyclases from bacteria, fungi, and plants hinder our understanding of the catalytic mechanism of LCY-Bs and CCS. In addition, the N terminus of plant LCY-B and CCS contains an amino sequence motif characteristic of a polypeptide predicted to adopt a Rossmann fold (Rossmann et al., 1974) and suggests the binding of an as yet unknown dinucleotide prosthetic ligand. It has been shown using recombinant bacterial enzyme that the cyclization of lycopene into β-carotene strictly requires NADPH but proceeds without any net redox change (Schnurr et al., 1996; Hornero-Mendez and Britton, 2002). Under the same conditions, FAD alone could not sustain bacterial LCY-B activity (Schnurr et al., 1996). Much less is known about the dinucleotide requirements of plant carotenoid cyclases, which are highly conserved within plants but are extremely divergent in nonplant organisms. Previously, a crucial acidic domain for lycopene cyclase activity was identified using an affinity-labeling strategy followed by site-directed mutagenesis (Bovier et al., 1997) in the absence of any crystal structures. This so-called 293-FLEET-297 motif of LCY-B and CCS contained two tandem Glu-295-Glu-296 residues that were essential for LCY-B- and κ-cyclase activities (Bovier et al., 1997). However, it still remains unclear how the protic mechanism is compatible with the requirement of dinucleotide cofactors.

To further explore the mechanism of plant carotenoid cyclases, we first choose pepper CCS as a prototypic enzyme because it displays a strong identity (52%) to pepper LCY-B, and we have shown previously that CCS could also catalyze the cyclization of lycopene into β-carotene (up to 25% of activity compared with LCY-B; Huguenev et al., 1995). Herein, we have shown that monomeric CCS purified to homogeneity from plant chromoplasts or recombinant CCS purified from Escherichia coli-transformed cells are typical flavoproteins containing one noncovalently bound FAD. We also observed that CCS-bound FAD is required for enzyme activity in the presence of NADPH, which functions as a reductant of FAD. During this process, no hydrogen is transferred to β-carotene or κ-cyclic carotenoids. In addition to this cofactor requirement, we also show from extensive site-directed mutagenesis using pepper CCS and LCY-B and Erwinia herbicola LCY-B (Mialoundama, 2009) that Glu-295 of pepper CCS and LCY-B plays a key role in the formation of β-carotene and κ-cyclic carotenoids, and we demonstrate that a similar role is played in structurally divergent bacterial LCY-Bs by Glu-196. These characteristics suggest that plant CCS and LCY-Bs are mechanistically similar to non-metal-assisted terpene cyclases, such as squalene:hopene cyclase and oxidosqualene cyclase, and additionally represent a new subfamily of flavoproteins like isopentenyl diphosphate isomerase type II, which catalyze carotenoid cyclization without any net redox modification of the substrate.

RESULTS

Identification of CCS as a Typical Flavoprotein

Amino acid sequence alignment indicates that plant-type LCY-Bs and CCS contain dinucleotide-binding motifs characteristic of an N-terminal Rossmann fold of FAD-dependent oxidoreductases (Rossmann et al., 1974; Cunningham et al., 1994; Huguenev et al., 1995). To analyze the nature of the potential dinucleotide prosthetic group, we noted during the initial purification steps that it was more tightly bound to CCS produced from pepper chromoplasts than that of recombinant CCS produced in E. coli. Based on this evidence, we first purified CCS from pepper chromoplasts using a previously described procedure (Bovier et al., 1994). When homogenous fractions (Fig. 1A) were subjected to mild electrophoretic conditions, the band corresponding to homogenous CCS protein could be observed without further treatment due to its yellow color, as observed previously (Bovier et al., 1994; Fig. 1B). Further analysis of chromoplast-derived CCS showed UV-visible absorption (458 nm) and fluorescence (520 nm) spectra characteristic of flavoproteins (Fig. 1C).

To further characterize the binding and the structure of the flavin prosthetic group, the yellow gel bands (Fig. 1B) were subjected to in-gel digestion using trypsin. The resulting peptide mixture was then extracted and analyzed by liquid chromatography-tandem mass spectrometry (LC-MS/MS). We could unambiguously identify CCS peptides, but no peptide fragments carrying covalently linked FAD or FMN were detected (data not shown). Subsequently, we extracted the yellow gel band using acidic acetonitrile before chromatographic analysis using FAD and FMN standards. Both cofactors provide the best signal in positive electrospray mode, and the best MS signal was obtained with 25 V (data not shown). The best specific multiple reaction monitoring (MRM) transitions were mass-to-charge ratio (m/z) 786 → 348 for FAD and 457 → 439 for FMN (Fig. 2). MRM analysis revealed the presence of FAD in the yellow gel extract (Fig. 2), whereas no FMN was detected (Fig. 2). The broadening of FAD and FMN peaks is probably due to the phosphate group, which is known to be poorly ionized under electrospray conditions. Collectively, these data demonstrate that FAD is noncovalently bound to chromoplast CCS (ChrCCS). The stoichiometry of 0.95 FAD per monomeric CCS could be determined at 458 nm, assuming that free and bound FAD have the same millimolar absorbance coefficient (11.3 mm⁻¹cm⁻¹; Whitby, 1953). Thus, one could postulate that each native CCS monomer (52 kD) contains one noncovalently bound FAD. Very similar data were obtained using purified recombinant pepper CCS (reCCS; Supplemental Fig. S1) and recombinant LCY-B (reLCY-B; data not shown). Binding affinities for FAD were determined for ChrCCS, reCCS, and reLCY-B. The dissociation constant values of 0.4 ± 0.02, 0.6 ± 0.03, and 0.9 ± 0.02 μM were obtained for ChrCCS, reCCS, and reLCY-B. These data collectively demon-
strate that CCS and plant LCY-B are FAD-binding flavoproteins.

Implication of FAD and NADPH during CCS Catalysis

To determine if FAD is implicated in the cyclization of lycopene into β-carotene or the formation of κ-cyclic carotenoids, we removed noncovalently bound FAD from ChrCCS and reCCS using hydrophobic interaction chromatography and determined the activity of the resulting apoenzymes under diverse reconstitution conditions. Apo-ChrCCS and apo-reCCS revealed no enzymatic activity toward carotenoid cyclization in the absence of added cofactors (Fig. 3A). Next, we analyzed the role of FAD following the reconstitution of apo-ChrCCS and apo-reCCS with FAD (Fig. 3A). The resulting holoenzyme catalyzed the cyclization of lycopene into β-carotene and the synthesis of κ-cyclic carotenoids only in the presence of FAD and NADPH (Fig. 3A). The NADPH requirement was also observed for purified ChrCCS and reCCS that contained noncovalently bound FAD in their native form. Similar data were obtained for rel.CY-B (Fig. 3B). The \( K_m \) values for NADPH were 0.15 ± 0.07, 0.25 ± 0.03, and 0.22 ± 0.02 mM for ChrCCS, reCCS, and rel.CY-B, respectively. Subsequently, we tested whether the requirement of FAD and NADPH is
coupled to hydrogen transfer to the cyclic carotenoids. To this end, we determined the CCS activity in the presence of FAD and (4R)-[4-3H]NADPH prepared as described previously (McCracken et al., 2004). Under these conditions, we did not observe tritium incorporation in the purified \( \beta \)-carotene or in capsanthin (data not shown). Thus, no hydrogen from NADPH is transferred to \( \beta \)-carotene and capsanthin, in agreement with previous data (Hornero-Mendez and Britton, 2002). Finally, we noted that upon incubation of the holoenzyme with NADPH, a strong modification of the absorption spectrum of holo-CCS was observed under anaerobic conditions (Fig. 3C). This change is reminiscent of the direct reduction of CCS-bound FAD, as noted for isopentenyl diphosphate isomerase type II, where noncovalently bound FMN could be reduced directly by NADH or NADPH (Hemmi et al., 2004; Rothman et al., 2007).

Implication of the Acidic FLEET Motif in CCS and LCY-B Activity

Based on the indirect implication of FAD and NADPH in the cyclization of carotenoid, we decided to explore further the role of amino acids potentially involved in the protic activation during carotenoid cyclization. A homology-based, structural alignment reveals that lycopene cyclases identified to date cannot be placed into a single homogenous group. The 293-FLEET-297 motif in LCY-Bs and in CCS (Supplemental Fig. S2) was previously shown to be crucial for cyclization of lycopene to \( \beta \)-carotene (Bouvier et al., 1997). Because neither crystal structure nor homology modeling is yet possible for LCY-B and CCS, we further analyzed the contribution of this motif in reCCS and reLCY-B (Mialoundama, 2009) by substituting Ala for targeted residues and by substituting Lys and Arg for selected acidic amino acids to convert them to positive charges. We also substituted Ala for conserved Asp-127, Asp-259, Glu-128, Glu-332, and His-360 residues, which are conserved between highly divergent plant and bacterial carotenoid cyclases (Supplemental Fig. S2). This strategy was used because Ala substitutions generate low distortions in protein structure (Fersht, 1987), while Lys and Arg substitutions eliminate the protic activation potential of acidic amino acid residues. Before mutant proteins were expressed in *E. coli*.
these substitutions were verified by DNA sequencing. Recombinant LCY-B and CCS were then purified by His-tag affinity chromatography as described in “Materials and Methods.” They migrated as a 66-kD band and were immunoreactive to antibodies directed against LCY-B or CCS (Fig. 4). The intensity of the protein signals as shown revealed that mutations of LCY-B and CCS have no effect on the expression or stability of proteins. Through the use of a FAD-agarose affinity column, we observed that the mutant and the wild-type reCCS and reLCY-B proteins had practically the same affinity for FAD (data not shown). The effects of different mutations on the cyclization of lycopene into β-carotene were compared with wild-type enzymes, and the kinetic properties were determined from Lineweaver-Burk plots (Lineweaver and Burk, 1934).

Substituting Ala, Lys, or Arg for Glu-295 completely abolishes reLCY-B (Table I) and reCCS (Table II) activity. Substituting Ala for Glu-296 reduces the lycopene cyclase activities of reLCY-B (Table I) and reCCS (Table II) to residual levels. Similarly, substituting Ala for Asp-127, Glu-128, Asp-259, and Glu-332 drastically reduced the activity up to 90%. In the case of reLCY-B H360A (Table I) or reCCS H360A (Table II), the activity was abolished but was partially recovered in reLCY-B H360K/R and reCCS H360K/R. Compared with the wild-type cyclases, the apparent V\textsubscript{max} values of the different mutant cyclases were reduced while the K\textsubscript{m} values were not altered (Tables I and II). Thus, the reduction of the catalytic activity was probably not due to a modified binding of lycopene. This indicates that the Asp-127, Glu-128, Asp-259, and His-360 residues of reLCY-B and reCCS have an important catalytic role but are not likely to directly initiate cyclization.

Examination of protein sequence data revealed that plant LCY-B or CCS Glu-295 is structurally equivalent to bacterial Glu-196 of Erwinia LCY-B (Fig. 5). This led us to envisage that this invariant residue may indicate the signal at approximately 66 kD.

<table>
<thead>
<tr>
<th>Enzyme</th>
<th>V\textsubscript{max}</th>
<th>K\textsubscript{m}</th>
<th>Relative Activity</th>
</tr>
</thead>
<tbody>
<tr>
<td>Wild type</td>
<td>130.0 ± 9</td>
<td>4.5 ± 0.03</td>
<td>100</td>
</tr>
<tr>
<td>D127A</td>
<td>75.0 ± 3</td>
<td>4.2 ± 0.04</td>
<td>57.6 ± 2</td>
</tr>
<tr>
<td>E128A</td>
<td>6.5 ± 0.3</td>
<td>4.0 ± 0.04</td>
<td>5.0 ± 0.2</td>
</tr>
<tr>
<td>D259A</td>
<td>32.5 ± 2.5</td>
<td>4.2 ± 0.02</td>
<td>25.0 ± 2</td>
</tr>
<tr>
<td>E295A</td>
<td>0</td>
<td>–</td>
<td>–</td>
</tr>
<tr>
<td>E295K</td>
<td>0</td>
<td>–</td>
<td>–</td>
</tr>
<tr>
<td>E295R</td>
<td>0</td>
<td>–</td>
<td>–</td>
</tr>
<tr>
<td>E296A</td>
<td>1.3 ± 0.04</td>
<td>ND</td>
<td>1.0 ± 0.04</td>
</tr>
<tr>
<td>E296K</td>
<td>0.3 ± 0.02</td>
<td>ND</td>
<td>0.2 ± 0.01</td>
</tr>
<tr>
<td>E296R</td>
<td>0.3 ± 0.02</td>
<td>ND</td>
<td>0.2 ± 0.01</td>
</tr>
<tr>
<td>E332A</td>
<td>6.5 ± 0.2</td>
<td>3.9 ± 0.03</td>
<td>5.0 ± 0.09</td>
</tr>
<tr>
<td>H360A</td>
<td>0</td>
<td>–</td>
<td>–</td>
</tr>
<tr>
<td>H360K</td>
<td>19.5 ± 1.5</td>
<td>5.0 ± 0.05</td>
<td>15.0 ± 0.8</td>
</tr>
<tr>
<td>H360R</td>
<td>26.0 ± 2</td>
<td>4.8 ± 0.07</td>
<td>20.0 ± 2</td>
</tr>
</tbody>
</table>

To further analyze the catalytic mechanism of plant carotenoid cyclase, we incubated wild-type and mutant reCCS with different 5,6-epoxy-xanthophylls and further analyzed the reaction products using ultra-performance liquid chromatography (UPLC)-MS/MS. When wild-type reCCS was incubated with antheraxanthin, we noted that, in addition to capsanthin, the normal product, a new xanthophyll (peak 1; Deli et al., 1998), was formed (Fig. 6A). The polarity, absorption maxima in the eluting solvent, and the total ion chromatogram of the new xanthophyll showed that when violaxanthin was used as a substrate, a new xanthophyll (peak 2) was formed (Fig. 6B). The polarity, absorption maxima in the eluting solvent, and the total ion chromatogram of the new xanthophyll (Supplemental Fig. S3) showed that it to be 5,6-diepikarpoxanthin (peak 1; Deli et al., 1998). In a similar vein, we incubated wild-type reCCS with violaxanthin. HPLC and MS analyses of reaction products showed that when violaxanthin was used as a substrate, a new xanthophyll (peak 2) was formed as expected and a new xanthophyll (peak 4) appeared as well (Fig. 6B). The relative elution of peak 4 on HPLC and its spectral characteristics in the eluting solvent are consistent with peak 4 being 5,6-diepilatoxanthin (Deli et al., 1998), but we could not further analyze its structure. The formation of these new xanthophylls was abolished when reCCSAla-295, reCCSLys-295, and reCCSH360A was expressed in E. coli.

**Figure 4.** Expression of plant-type LCY-Bs. Soluble lysate from cells harboring the control vector and lines expressing wild or mutated plant reLCY-B or reCCS was used for affinity purification as described in “Materials and Methods.” Affinity-purified wild and mutant reLCY-B or reCCS were probed with anti-LCY-B or anti-CCS. The arrowheads indicate the signal at approximately 66 kD.
DISCUSSION

FAD and NADPH Dependence of Plant Carotenoid Cyclases

Amino acid sequence alignment of plant-type LCY-Bs suggests that the N-terminal region could adopt a Rossmann-fold characteristic of flavoproteins (Dym and Eisenberg, 2001). The nature of the putative prosthetic flavin group had remained elusive. Previously, a strict requirement of exogenous NADPH, and nonessentially, of exogenous FAD, have been reported for the enzymatic cyclization of lycopene into \( \beta \)-carotene using \textit{Erwinia uredovora} LCY-B (Schnurr et al., 1996). However, although \textit{E. uredovora} LCY-B functions in the presence of NADPH, no hydrogen is transferred from NADPH to \( \beta \)-carotene (Hornero-Mendez and Britton, 2002). This reinforced the accepted protic cyclization mechanism of carotenoids triggered by unknown amino acid residues but leaves unexplained the indirect role played by NADPH. Using CCS as a carotenoid cyclase prototype has allowed us to identify nonbound FAD as the prosthetic group implicated in the synthesis of \( \beta \)-carotene and \( \kappa \)-cyclic carotenoids. Our results also reveal that the cyclization reaction takes place only in the presence of FAD and NADPH (Fig. 3, A and B). Data further revealed that when NADPH is incubated with CCS holoenzyme, its absorption spectrum is drastically modified (Fig. 3C). The characteristic absorption at approximately 450 nm was lost (Fig. 3C). A similar change was observed when dithionite was used instead of NADPH (data not shown). The spectral change suggests a direct reduction of FAD by NADPH (Fig. 3C). In \textit{E. uredovora}, FAD reduced by Ti(III)-citrate in the absence of NADPH could sustain the conversion of lycopene into \( \beta \)-carotene (Yu et al., 2010; P. Beyer, personal communication). Very similar observations have been described for isopentenyl diphosphate isomerase type II (Hemmi et al., 2004; Rothman et al., 2007), for which the reaction requires the reduction of FMN-bound enzyme by NADPH. In this context, it is interesting that the characteristics of CCS are reminiscent of those described for isopentenyl diphosphate isomerase type II, which is a flavoenzyme, and requires FMN and NADPH for activity but functions without net redox modifications of the substrate. Because flavoproteins usually catalyze redox reactions, the question arises of how CCS-bound FAD functions. With respect to data obtained from isopentenyl diphosphate isomerase type II, which has been crystallized, two likely explanations could be offered (Rothman et al., 2008; Unno et al., 2009). In one case, CCS-bound FAD, when reduced by NADPH, could act as an acid-base catalyst (Unno et al., 2009). Alternatively, reduced CCS-bound FAD could be involved in the stabilization of the transient carbocation formed during the reaction (Rothman et al., 2008). The latter conclusion is sup-

![Figure 5](https://example.com/figure5.png)

**Figure 5.** Functional characterization of bacterial wild-type and mutant LCY-Bs. Following the identification of Glu-196 as an active-site residue of \textit{Erwinia} lycopene cyclase, a plasmid-based assay was used to test wild-type and mutant \textit{Erwinia} LCY-B. \textit{E. coli} JM101 harboring the plasmid pACYC-EBI (chloramphenicol), which produces lycopene, was used. After transformation of \textit{E. coli} JM101-pACYC-EBI (chloramphenicol) with plasmids encoding wild-type or mutant \textit{Erwinia} LCY-Bs and induction with 0.5 mm isopropyl-\( \beta \)-thiogalactopyranoside, total carotenoids were extracted from the bacterial pellets and analyzed by HPLC as described in “Materials and Methods.” Data shown represent means \pm s for two independent experiments.
ported by the recent finding that \textit{E. uredovora} LCY-B could function with reduced 1- and 5-deazaflavin analogs that probably have limited potential to trigger the protic activation of lycopene (Yu et al., 2010).

**Figure 6.** HPLC analysis of \(k\)-cyclic carotenoids synthesized by reCCS from xanthophyll epoxides. A, Anthaxanthin conversion into \(k\)-cyclic and 3,5,6-trihydroxy xanthophyll by reCCS. ReCCS was incubated with anthaxanthin as described in “Materials and Methods,” and products were analyzed by HPLC and online diode array detection. Peaks refer to 5,6-diepikarpoxanthin (peak 1), capsanthin (peak 2), and anthaxanthin (peak 3). B, Violaaxanthin conversion into \(k\)-cyclic and 3,5,6-trihydroxy xanthophyll by reCCS. ReCCS was incubated with violaxanthin, and products were analyzed as described in A. Peaks refer to putative 5,6-diepilatoxanthin (peak 4), capsorubin (peak 5), capsanthin-5,6-epoxide (peak 6), and violaxanthin (peak 7).

**Importance of Invariant Acidic Residues in Plant Carotenoid Cyclases**

LCY-Bs cloned from bacteria, plants, and fungi do not contain consensus motifs. We have previously used a site-directed mutant approach to identify several amino acid residues critical for differential inhibition of plant LCY-B and CCS (Bouvier et al., 1997). In this study, we made use of an Ala scanning mutagenesis strategy combined with the introduction of positive or negative residues to replace several conserved residues between plant cyclases LCY-B, CCS, and bacterial LCY-B. Stably expressed recombinant proteins were purified and used for enzyme assays. The mutant proteins allowed us to define Glu-295 as a critical residue for the cyclization of lycopene into \(\beta\)-carotene and also for the formation of \(k\)-cyclic carotenoids in plants. Glu-295A/K/R failed to synthesize \(\beta\)-carotene or \(k\)-cyclic carotenoids. In addition, the mutation of Glu-196 of \textit{E. herbicola} (Fig. 5) or \textit{E. uredovora} (Yu et al., 2010; P. Beyer, personal communica-}

...tion), equivalent to Glu-295 of plant-type LCY-Bs, abolished the cyclization of lycopene into \(\beta\)-carotene. Therefore, the presence of a Glu residue at position 295 in plant enzymes or at position 196 in bacteria is essential for the catalytic activity of carotenoid cyclases. Cloning of fungal LCY-B from wild-type and mutant strains of \textit{Phycomyces blakesleeanus} (Arrach et al., 2001) supports our conclusion. Fungal LCY-Bs have no homology with plant cyclases; however, massive amounts of lycopene accumulated in the \textit{Phycomyces} Car23 mutant, which has a Glu-77-to-Lys-77 substitution. Glu-77 represents the acidic residue in the PxE(E/D) motif found in LCY-Bs of different organisms, including \textit{Mycobacterium aurum}, \textit{Sulfolobus solfataricus}, and fungi. Thus, although bacterial, fungal, and plant-type LCY-Bs are highly divergent, one could suggest that they have evolved to convert lycopene into \(\beta\)-carotene via an acid-catalyzed cyclization mechanism. Indeed, the formation of 5,6-diepikarpoxanthin and of putative 5,6-diepilatoxanthin catalyzed by CCS (Fig. 6) could be rationalized by considering that following the protic attack, the incipient carotenoid carbocation generated is flexibly stabilized to yield different structures (Fig. 7). This has been hypothesized previously (Deli et al., 1998), because these new xanthophylls are present in trace amounts in red pepper fruits and absent in yellow varieties, which are known to be devoid of CCS (Bouvier et al., 1994).

**Figure 7.** Molecular diversity of carotenoids generated by CCS. A, Flexible stabilization of incipient carbocation generated by CCS. B, Possible pathways of carotenoids produced by CCS during the formation of \(k\)-cyclic carotenoids.
Proton activations of carbon-carbon double bonds are uncommon in nature. Typical examples are Δ8-Δ7 cholesterol isomerase (Wilton et al., 1969), isopenstenyldiphosphate isomerase (Street et al., 1990), and squalene cyclase (Wendt et al., 2000). While Cys-67 has been identified as the protic activator in isopenstenyldiphosphate isomerase (Street et al., 1990), Asp residues are the proton initiators in squalene cyclase. For squalene cyclase, protonation of the terminal double bond of squalene is triggered by Asp-376 of the D376DTAVV motif and assisted by His-451 (Wendt et al., 2000). It has been suggested that the tandem aspartic residues in the DDTAVV motif of squalene cyclase offer higher acidity, which is required for protonating the terminal double bond of squalene compared with the terminal oxirane of oxidosqualene, the phytosterol precursor, which is more easily protonated (Wendt et al., 2000). It is possible that the tandem Glu residues in the FLEET motif of LCY-Bs and CCS can play the same role. Interestingly, the similarity between the catalytic mechanism of squalene cyclase and LCY-B or CCS is supported by the fact that aza-squalene, a potent transient-state inhibitor of squalene cyclase (Reinert et al., 2004), is also a strong inhibitor of pepper LCY-B and CCS-(Camaras et al., 1985; Bouvier et al., 1997). Due to these characteristics, and because flavin prosthetic groups are not involved in squalene cyclase catalysis, one may suggest that CCS- or plant LCY-B-bound FAD, in its reduced form, is probably involved in the stabilization of the incipient carotenoid carboxation (Yu et al., 2010) and could function as previously envisioned “negative point charges” (Bouvier et al., 1997). Given previous biochemical data (Schnurr et al., 1996; Hornero-Mendez and Britton, 2002) indicating the requirement of NADPH for LCY-B activity, we show that FAD is likely induced by NADPH. Further studies are required to fully evaluate this hypothesis.

In conclusion, the data presented herein reveal that plant-type carotenoid cyclases have evolved to integrate two characteristics. First, they belong to FAD-dependent flavoproteins that catalyze reactions without net redox changes (Bornemann, 2002). In the terpene series, this feature is shared with isopensteyldiphosphate isomerase II. Second, they belong to the class of non-metal-coordinated cyclases such as squalene and oxidosqualene cyclases that operate via a protic attack. With respect to the latter mechanism, we show that Glu-295 of plant-type carotenoid cyclases have evolved to integrate the catalytic feature is shared with isopensteyldiphosphate isomerase (Street et al., 1990). Assp residues are the proton initiators in squalene cyclase. For squalene cyclase, protonation of the terminal double bond of squalene is triggered by Asp-376 of the D376DTAVV motif and assisted by His-451 (Wendt et al., 2000). It has been suggested that the tandem aspartic residues in the DDTAVV motif of squalene cyclase offer higher acidity, which is required for protonating the terminal double bond of squalene compared with the terminal oxirane of oxidosqualene, the phytosterol precursor, which is more easily protonated (Wendt et al., 2000). It is possible that the tandem Glu residues in the FLEET motif of LCY-Bs and CCS can play the same role. Interestingly, the similarity between the catalytic mechanism of squalene cyclase and LCY-B or CCS is supported by the fact that aza-squalene, a potent transient-state inhibitor of squalene cyclase (Reinert et al., 2004), is also a strong inhibitor of pepper LCY-B and CCS-(Camaras et al., 1985; Bouvier et al., 1997). Due to these characteristics, and because flavin prosthetic groups are not involved in squalene cyclase catalysis, one may suggest that CCS- or plant LCY-B-bound FAD, in its reduced form, is probably involved in the stabilization of the incipient carotenoid carboxation (Yu et al., 2010) and could function as previously envisioned “negative point charges” (Bouvier et al., 1997). Given previous biochemical data (Schnurr et al., 1996; Hornero-Mendez and Britton, 2002) indicating the requirement of NADPH for LCY-B activity, we show that FAD is likely induced by NADPH. Further studies are required to fully evaluate this hypothesis.

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and separation at 6°C, we first adopted an in-gel digestion procedure followed by MS analysis using an automated protein digestion system (MassPREP Station; Waters) as described previously (Heintz et al., 2009) to determine whether the prosthetic group was covalently bound. The gel was subsequently extracted using acetonitrile acidified with 1% formic acid. The resulting yellow supernatant was clarified by centrifugation, dried under argon, and solubilized in 200 μL of water:acetonitrile (90:10, v/v) before UPLC analysis on an Acquity UPLC system (Waters), using authentic FAD and FMN standards. The column was eluted with water containing 0.1% formic acid (solvent A) and acetonitrile containing 0.1% formic acid (solvent B). The flow rate was 0.40 mL min⁻¹. Chromatographic conditions were as follows: linear gradient from 95% A to 5% B (2 min); 95% A to 100% B (6 min); 100% B (2 min); 100% B to 95% A (1 min); 95% A to 100% A (2 min). UV-visible spectra were recorded from 200 to 500 nm. The electrospray ionization source was used in positive and negative mode with capillary voltage of 3.4 kV; radiofrequency lens at 0 V, resolution (L1, H1, L2, H2) 15, ion energy 1 and 2: 0.5. Source and desolvation temperatures were 135°C and 400°C. MRm with positive electrospray ionization was used for qualitative analysis. Cone tensions and collision energy were optimized at 25 and 20 V for each compound. For FAD, the MRm transition was m/z 785.8–348, and for FMN, the MRm transition was m/z 457–439. The prosthetic group of reCCS or reLCY-B was extracted and analyzed using the same procedure. Purified CCS in phosphate buffer (pH 7) containing 10% glycerol and 0.5% Tween 80 was used for UV-visible absorption spectrophotometry using a NanoDrop 2000 (Thermo Scientific) or a Shimadzu UV2401 spectrophotometer, and fluorescence emission spectra were recorded with a Shimadzu RF5302 fluorescence spectrophotometer at room temperature following excitation at 458 nm. The flavin content was determined as described previously (Whitby, 1953).

Deflavinylation and Reconstitution of Carotenoid Cyclases

To prepare the apoenzyme of carotenoid cyclases, we modified a previously described procedure (Van Berkel et al., 1988). Purified carotenoid cyclases were adsorbed onto a PhenylSepharose column (1 × 5 cm) equilibrated with 50 mM K₂HPO₄ buffer (pH 7) containing 1 mM EDTA, 2 mM NaCl, and 1 mM KBr. FAD was released using the same buffer adjusted to pH 5. The apoproteins were subsequently eluted with 50 mM Tris-maleate buffer (pH 6.8) containing 15% glycerol and 40% dimethyl sulfoxide (DMSO). DMSO was eliminated by dialysis against the same buffer devoid of DMSO, and the fluorescence of the apoproteins was recorded at 520 nm following excitation at 458 nm. The flavin content was determined using GraphPad Prism 4.0 software.

Carotenoid Cyclase Assay

Unless otherwise stated, assays were performed in a reaction mixture (final volume of 250 μL) containing 50 mM Tris-maleate buffer, pH 6.8, 1 mM diithiothreitol, 2 mM NADPH, 10 μM FAD, 1 mg of Tween 80, 50 mMol of carotenoid substrates, and a definite amount of enzyme. The reaction mixture was flushed with argon and incubated at 25°C for 30 min to 1 h. The reaction products were isolated following the addition of 500 μL of chloroform/methanol (2:1, v/v) before HPLC analysis of the lipid phase (Camara, 1985). General methods, including carotenoid analysis and identification, were as described previously (Davies, 1976). The HPLC system consisted of SpectraSYSTEM (ThermoFinnigan) comprising an SCM1000 solvent degasser, a P1-1000 gradient pump, an AS3000 autosampler, and a UV/VisDLPI diode array detector. Data acquisition and processing were done using ChromQuest version 3 software (ThermoFinnigan). The reaction products were collected, evaporated to dryness, and further characterized by UPLC-MS/MS using the MS mode. The analysis was performed on a Waters Quattro Premier XE equipped with an atmospheric pressure chemical ionization (APCI) source and coupled to an Acquity UPLC system (Waters) with diode array detector. UV spectra were recorded from 200 to 500 nm. Chromatographic separation was achieved using an Acquity UPLC BEH C₁₈ column (100 × 2.1 mm, 1.7 μm; Waters) coupled to an Acquity UPLC BEH C₁₈ precolumn (2.0 × 5 mm, 1.7 μm; Waters). The mobile phases were 75% methanol in water (solvent A) and 100% isopropanol (solvent B). The following separation sequence was adopted: 100% A (5 min); 100% A to 50% B (3 min); 50% A + 50% B (4 min); 50% A to 100% B (5 min); 100% B (10 min); 100% B to 50% A (2 min); 50% B + 50% A (5 min); 50% B to 100% A (6 min). The parameters for MS detection and APCI were as follows: nebulizer gas flow was set to approximately 50 L h⁻¹, and the desolvation gas flow was 150 L h⁻¹. The API photoelectric response was set to 500°C, and the source temperature was 120°C. The capillary voltage was set at 1.5 kV, and cone voltage after optimization for each carotenoid was set to 25 V. The ionization mode was positive. Low-mass and high-mass resolution were 15 for both mass analyzers, ion energies 1 and 2 were 1 V, entrance and exit potentials were 50 V, and detector (multiplier) gain was 650 V. Mass spectra of carotenoids were acquired with a scan range of m/z 350 to 750 atomic mass units. The combination of the chromatographic retention time (tₚ) and the parent mass was used to selectively monitor the different carotenoids as follows: antheraxanthin ([M+H⁺] = 586, tₚ = 8.47), violaxanthin ([M+H⁺] = 602, tₚ = 8.13), capsanthin ([M+H⁺] = 586, tₚ = 8.32), capsanthin-5,6-epoxide ([M+H⁺] = 602, tₚ = 7.96), capsorubin ([M+H⁺] = 584, tₚ = 7.81), and 5,6-diepi-karotenoid ([M+H⁺] = 602, tₚ = 7.59). Data acquisition and analysis were performed with Mass Lynx software (version 4.1) running under Windows XP professional on a Pentium personal computer. The kinetic parameters were obtained by nonlinear regression analysis using the GraphPad Prism 4.0 program (GraphPad Software).

Sequence data from this article can be found in the GenBank/EMBL data libraries under accession numbers CAAS60119 for Capsicum LCY-B and CAAS37359 for Capsicum CCS.

Supplemental Data

The following materials are available in the online version of this article.

Supplemental Figure S1. Spectral analysis of purified reCCS.

Supplemental Figure S2. Multiple alignment of deduced amino acid sequences of selected LCYs and pepper CCS.

Supplemental Figure S3. Total ion chromatogram analysis of carotenoid products isolated from the incubation of reCCS with xanthophyll epoxides.

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LITERATURE CITED


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